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Exergy - Wikipedia

In thermodynamics, the exergy of a system is the maximum useful work possible during a process that brings the system into equilibrium with a heat reservoir, reaching maximum entropy. When the surroundings are the reservoir, exergy is the potential of a system to cause a change as it achieves equilibrium with its environment. **What Is Exergy? Exergy Economics**

Exergy is a thermodynamic concept, used for many years within engineering analyses of chemical and mechanical processes and systems. Officially, exergy is defined as: The maximum useful work which can be extracted from a system as it reversibly comes into equilibrium with its environment. In other words, it is the capacity of energy to do

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This in-depth discussion proposes what Exergy will accomplish and how it can be utilized, by illustrating how its technical value relates to the services that consumers and third parties can provide to one another and to the electric power system.

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Exergy | Article about Exergy by The Free Dictionary **exergy** [eks' rj] (thermodynamics) The portion of the total energy of a system that is available for conversion to useful work; in particular, the quantity of work that can be performed by a fluid relative to a reference condition, usually the surrounding ambient condition. Exergy a term used in thermodynamics to designate the maximum

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