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[Atomie nucleus - Nuclear Energy](#)
 Properties of the atomic nucleus Almost the entire mass of an atom is found in the atomic nucleus, with a very small contribution from the electron cloud because electrons weigh very little compared to neutrons and protons .
[The Structure and Properties of Atomic Nuclei | SpringerLink](#)

Its general theme was the Structure and Properties of Atomic Nuclei . This was the last conference to take place before World War II. A conference on elementary particles and their mutual interactions had been planned for 1939, but world events intervened and it had to be cancelled.
[Chapter 8. The Atomic Nucleus - Western University](#)
[Chapter 8: The Atomic Nucleus Notes: Most of the material in this chapter is taken from Thornton and Rex, Chapters 12 and 13.](#)
[8.1 Nuclear Properties](#) Atomic nuclei are composed of protons and neutrons, which are referred to as nucleons. Although both types of particles are not fundamental or elementary, they can still be considered as basic constituents for the purpose of understanding the Basic properties of atomic nuclei - Warsaw University of ...

Basic properties of atomic nuclei radius of a nucleus $3.1 R = R_0 A$ where R_0 is an experimentally determined constant: $R = 1.2 \cdot 10^{-15} \text{ m} = 1.2 \text{ fm}$ $0 = = A$ is called the nucleon number.

[BibTeX bibliography fermi-enrico.bib](#)
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[Atomic Nuclei Explain How Supernovae Formed Heavier ...](#)
 Atomic nuclei consist of a good mix of protons and neutrons. The definition of an element is determined by the number of protons. The physical characteristics of elements depend also on the number of neutrons. The various states are called isotopes.

[Atomic Nucleus: Definition, Structure & Size - Video ...](#)
 The atomic nucleus is the central part of the atom. There is a lot to be told by the structure of the atomic nucleus. This

lesson goes through the structure of the atomic nucleus and other factors

XIII. Properties of Atomic Nuclei - ScienceDirect

This chapter reviews a work of Niels Bohr on properties of atomic nuclei. The usual physical and chemical properties of materials are concerned with changes in the outer electron system radioactive phenomena in certain elements are caused by processes in the nucleus itself.